

5-(Furan-2-yl)-4-(*p*-tolyl)-2,4-dihydro-1,2,4-triazole-3-thioneNecmi Dege,<sup>a\*</sup> Omer Andac,<sup>b</sup>  
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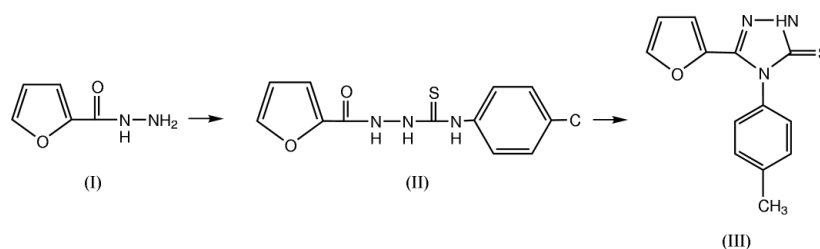
## Key indicators

Single-crystal X-ray study  
*T* = 293 K  
Mean  $\sigma$ (C–C) = 0.005 Å  
*R* factor = 0.058  
*wR* factor = 0.110  
Data-to-parameter ratio = 15.4For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

In the title molecule, C<sub>13</sub>H<sub>11</sub>N<sub>3</sub>OS, the triazole plane forms dihedral angles of 3.8 (2) and 81.3 (2)° with the furan and *p*-tolyl ring planes, respectively. In the crystal structure, the molecules exist as centrosymmetric N–H···S hydrogen-bonded dimers, with an N···S distance of 3.312 (2) Å. The packing is further stabilized by C–H···O, C–H···N, C–H··· $\pi$  and  $\pi$ – $\pi$  interactions.

## Comment

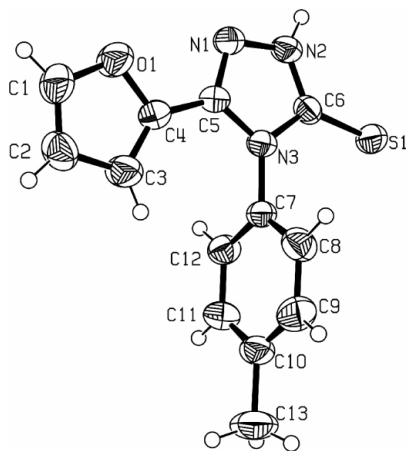
Derivatives of 1,2,4-triazole are known to exhibit anti-inflammatory (Unangst *et al.*, 1992; Mullican *et al.*, 1993), antiviral (Jones *et al.*, 1965), analgesic (Sughen & Yoloye, 1978), antimicrobial (Shams El-Dine & Hazzaa, 1974; Misato *et al.*, 1977; Cansız *et al.*, 2001), anticonvulsant (Stillings *et al.*, 1986) and antidepressant activities (Kane *et al.*, 1988), the latter being usually explored by the forced swim test (Porsolt *et al.*, 1977; Vamvakides, 1990). Among the pharmacological profiles of 1,2,4-triazoles, their antimicrobial, anticonvulsant and antidepressant properties seem to be the best documented. The title compound, (III), has been newly synthesized by the reaction of *p*-tolyl isothiocyanate and furan-2-carboxylic acid hydrazide, (I), through 1-(2-furoyl)-4-*p*-tolylthiosemicarbazide, (II) (see scheme). Base-catalysed intramolecular dehydrative cyclization of this intermediate furnished the thione (III) in good yield (79%) (Cansız *et al.*, 2003); its crystal structure is presented here.



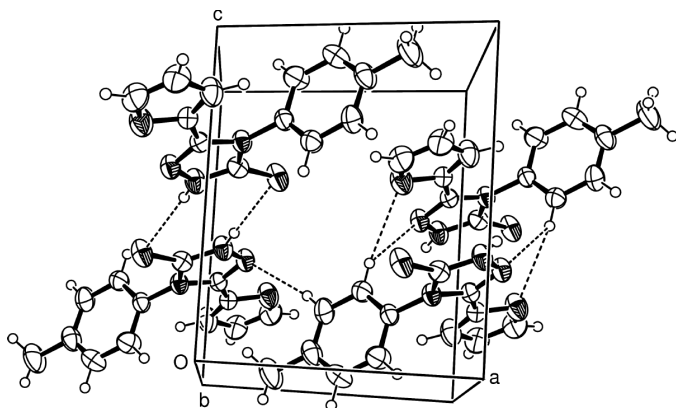
The molecular structure of (III) is non-planar (Fig. 1). The triazole plane forms dihedral angles of 3.8 (2) and 81.3 (2)° with the furan and *p*-tolyl ring planes, respectively [C3–C4–C5–N3 = –5.3 (6)°, O1–C4–C5–N1 = –3.0 (4)°, C5–N3–C7–C12 = 79.3 (4)° and C6–N3–C7–C8 = 83.9 (4)°].

N–H···S, C–H···O and C–H···N intermolecular hydrogen bonds are observed in the crystal structure (Table 2 and Fig. 2). The N2–H2A···S1<sup>i</sup> hydrogen bond links inversion-related molecules into dimers. The N···S distance [3.312 (2) Å] in this interaction is shorter than the mean value of 3.44 (1) Å reported for such hydrogen bonds by Allen *et al.* (1997); also, the N–H···S angle (164°) is wider than the mean angle of 158 (1)°. By comparing the N–H···S hydrogen bonding in *N*-benzoyl-*N'*-methyl-*N'*-phenylthiourea, (IV), *N*-

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**Figure 1**  
An ORTEP-3 (Farrugia, 1997) drawing of (III), with the atom-numbering scheme. Displacement ellipsoids for non-H atoms are drawn at the 50% probability level. H atoms are shown as spheres of arbitrary radii.



**Figure 2**  
A packing diagram of (III), illustrating the hydrogen-bonding network (dashed lines).

benzoyl-*N'*-(3,4-dimethylphenyl)thiourea, (V) (Shanmuga Sundara Raj *et al.*, 1999), 5-(furan-2-yl)-1,3,4-oxadiazole-2(3*H*)-thione, (VI) (Öztürk, Akkurt, Cansız, Çetin *et al.*, 2004), and 4-(4-chlorophenyl)-3-(furan-2-yl)-1*H*-1,2,4-triazole-5(4*H*)-thione, (VII) (Öztürk, Akkurt, Cansız, Koparır *et al.*, 2004), it can be seen that dimer formation shortens the N...S distances. The N...S distances in (III), (IV), (V), (VI) and (VII), which exist as centrosymmetric N—H...S hydrogen-bonded dimers, are 3.312 (2), 3.473 (1), 3.501 (2), 3.321 (3) and 3.304 (2) Å, respectively. The C12—H12...O1<sup>iii</sup> and C12—H12...N1<sup>iii</sup> weak interactions (Desiraju, 1996) constitute a pair of bifurcated hydrogen bonds [symmetry code: (iii)  $-x, -y, 1 - z$ ; Fig. 2].

A PLATON analysis (Spek, 1997) showed the presence of a weak  $\pi$ – $\pi$  interaction between the furan ring and triazole ring of the inversion-related molecule at  $(-x, -y, 1 - z)$  [centroid–centroid distance is 3.611 (2) Å]. Furthermore, C—H... $\pi$  interactions involving the benzene ring are also observed in the crystal structure (Table 2).

## Experimental

A mixture of furan-2-carboxylic acid hydrazide, (I) (0.01 mol), and *p*-tolyl isothiocyanate (0.01 mol) in dry C<sub>6</sub>H<sub>6</sub> was refluxed for 6 h. The

solid material obtained on cooling was filtered off and recrystallized from methanol to yield 1-(2-furoyl)-4-*p*-tolylthiosemicarbazide, (II). A stirred mixture of compound (II) (1 mmol) and sodium hydroxide (40 mg, 1 mmol, as a 2 N solution) was refluxed for 4 h. After cooling, the solution was acidified with hydrochloric acid and the precipitate was filtered off to obtain the title compound, (III). The precipitate was then crystallized from ethanol (yield 79%, m.p. 530–551 K). IR  $\nu$  (cm<sup>-1</sup>): 3351, 3285 (N—H), 2576 (SH), 1621 (C=N), 1250 (C—O—C), 1534, 1258, 1050, 951 (N—C=S, amide I, II, III and IV bands). <sup>1</sup>H NMR:  $\delta$  2.46 (*s*, 3H, CH<sub>3</sub>), 5.94–6.36 (*m*, 3H, furan), 7.31–7.47 (*m*, 4H, Ar-H), 14.01 (*s*, 1H, SH, or NH). Calculated for C<sub>13</sub>H<sub>11</sub>N<sub>3</sub>OS: C 60.68, H 4.31, N 16.33, S 12.46%; found: C 60.65, H 4.39, N 16.50, S 12.42%.

## Crystal data

|   |   |
|---|---|
| C <sub>13</sub> H <sub>11</sub> N <sub>3</sub> OS | Z = 2   |
| <i>M<sub>r</sub></i> = 257.31                     | <i>D<sub>x</sub></i> = 1.332 Mg m <sup>-3</sup> |
| Triclinic, <i>P</i> $\bar{1}$                     | Mo <i>K</i> $\alpha$ radiation                  |
| <i>a</i> = 8.5665 (15) Å                          | Cell parameters from 7019 reflections           |
| <i>b</i> = 8.7759 (16) Å                          | $\theta$ = 2.1–27°                              |
| <i>c</i> = 9.5372 (17) Å                          | $\mu$ = 0.24 mm <sup>-1</sup>                   |
| $\alpha$ = 91.796 (15)°                           | <i>T</i> = 293 (2) K                            |
| $\beta$ = 91.407 (14)°                            | Prism, colorless                                |
| $\gamma$ = 116.386 (12)°                          | 0.31 × 0.20 × 0.07 mm                           |
| <i>V</i> = 641.4 (2) Å <sup>3</sup>               |   |

## Data collection

|  |  |
|--|--|
| Stoe IPDS-2 diffractometer                                       | 2518 independent reflections           |
| $\omega$ scans   | 1539 reflections with $I > 2\sigma(I)$ |
| Absorption correction: by integration ( <i>X-RED32</i> ;         | <i>R</i> <sub>int</sub> = 0.090        |
| Stoe & Cie, 2002)  | $\theta_{\max}$ = 26.0°                |
| <i>T</i> <sub>min</sub> = 0.937, <i>T</i> <sub>max</sub> = 0.987 | <i>h</i> = -10 → 10                    |
| 8282 measured reflections  | <i>k</i> = -10 → 10                    |
|  | <i>l</i> = -11 → 11                    |

## Refinement

|                                     |  |
|-------------------------------------|--|
| Refinement on <i>F</i> <sup>2</sup> | H-atom parameters constrained                        |
| $R[F^2 > 2\sigma(F^2)] = 0.058$     | $w = 1/[\sigma^2(F_o^2) + (0.0375P)^2]$              |
| $wR(F^2) = 0.110$                   | where $P = (F_o^2 + 2F_c^2)/3$                       |
| <i>S</i> = 1.01                     | ( $\Delta/\sigma$ ) <sub>max</sub> = 0.020           |
| 2518 reflections                    | $\Delta\rho_{\max} = 0.21 \text{ e } \text{Å}^{-3}$  |
| 164 parameters                      | $\Delta\rho_{\min} = -0.21 \text{ e } \text{Å}^{-3}$ |

**Table 1**

Selected geometric parameters (Å, °).

|          |           |          |           |
|----------|-----------|----------|-----------|
| S1—C6    | 1.675 (3) | N3—C7    | 1.448 (3) |
| N1—C5    | 1.298 (3) | C5—N3    | 1.378 (4) |
| N2—C6    | 1.332 (4) | C6—N3    | 1.380 (3) |
| N2—N1    | 1.366 (3) |          |           |
| C1—O1—C4 | 106.5 (2) | N1—C5—N3 | 111.0 (2) |
| C5—N1—N2 | 104.1 (2) | N1—C5—C4 | 123.6 (3) |
| C6—N2—N1 | 114.0 (2) | N3—C5—C4 | 125.5 (2) |
| C5—N3—C6 | 107.6 (2) | N2—C6—N3 | 103.2 (2) |
| C5—N3—C7 | 127.2 (2) | N2—C6—S1 | 129.2 (2) |
| C6—N3—C7 | 125.0 (2) | N3—C6—S1 | 127.6 (2) |

**Table 2**

Hydrogen-bonding geometry (Å, °).

| <i>D</i> —H... <i>A</i>     | <i>D</i> —H | H... <i>A</i> | <i>D</i> ... <i>A</i> | <i>D</i> —H... <i>A</i> |
|-----------------------------|-------------|---------------|-----------------------|-------------------------|
| N2—H2A...S1 <sup>i</sup>    | 0.86        | 2.48          | 3.312 (2)             | 164                     |
| C11—H11...N1 <sup>ii</sup>  | 0.93        | 2.56          | 3.468 (4)             | 165                     |
| C12—H12...O1 <sup>iii</sup> | 0.93        | 2.75          | 3.567 (4)             | 147                     |
| C12—H12...N1 <sup>iii</sup> | 0.93        | 2.69          | 3.477 (4)             | 143                     |
| C1—H1...CgP <sup>iv</sup>   | 0.93        | 2.85          | 3.663 (4)             | 146                     |
| C3—H3...CgP                 | 0.93        | 2.96          | 3.706 (4)             | 138                     |

Symmetry codes: (i)  $-x, 1 - y, 1 - z$ ; (ii)  $1 + x, y, z$ ; (iii)  $-x, -y, 1 - z$ ; (iv)  $x - 1, y - 1, z$ ; CgP is the centroid of the benzene ring.

All H atoms were placed in geometrically idealized positions and allowed to ride on their parent atoms, with O—H, N—H and C—H distances of 0.82, 0.86 and 0.93 Å, respectively. The  $U_{\text{iso}}(\text{H})$  values were set equal to  $1.5U_{\text{eq}}(\text{O})$  for the hydroxy and ethane H atoms, and  $1.2U_{\text{eq}}(\text{parent atom})$  for the other H atoms.

Data collection: *X-AREA* (Stoe & Cie, 2002); cell refinement: *X-AREA*; data reduction: *X-RED32* (Stoe & Cie, 2002); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997) and *PLATON* (Spek, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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